

Risk Assessment of Disinfection By-Product Formation

Henry MACKEOWN – *PhD student, Lille 1 University Supervisor* – Professor Baghdad OUDDANE, *Lille 1 University Co-supervisor* – University Lecturer Justine CRIQUET, *Lille 1 University*















Risk assessment of Disinfection By-Product ^{1/17} **formation**

Outline

- I. Context
- II. Research interest
- III. Literature Review
- IV. Development of analytical methods
- V. First results
- VI. Upcoming work















Minimising Disinfection By-Products

- Strategy information and local specificities for partners and observers for Organic Matter treatment" (Irene Caltran, Delft)
 - DBP minimisation as one of the main reasons for improving DOC removal

Improving Dissolved Organic Carbon **removal** to **reduce** Disinfection By-Product formation Developing a **risk assessment** of **DBP formation** in the 2 seas area, looking at the **benefits** of **DOC removal** on DBP formation and speciation









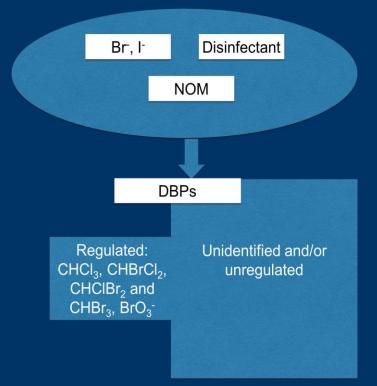






Disinfection By-Product characterization

- Oxidation of Natural Organic Matter by a disinfectant
- HOCI/CIO⁻ + NOM \rightarrow CI-DBPs
- HOCI/CIO⁻ + Br⁻ \rightarrow HOBr/BrO⁻ +CI⁻
- HOCI/CIO⁻ + I⁻ \rightarrow HOI/IO⁻ +CI⁻
- HOBr/BrO⁻ + NOM \rightarrow Br-DBPs
- HOI/IO⁻ + NOM \rightarrow I-DBPs











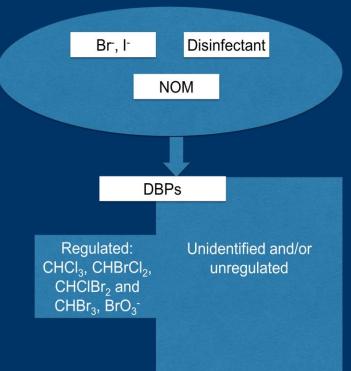






Disinfection By-Product regulations

- Sum of 4 Trihalomethanes: 100 μg/L (EU), 80 μg/L (US), 250 μg/L (Australia)
- Sum of 5 Haloacetic Acids: 60 μg/L (US)
- Several chlorophenols and chloroacetic acids (Australia)
- Bromate : 10 μg/L (EU and US), 80 μg/L (Australia)
- Chlorite: 1 mg/L (US), 800 μg/L (Australia)

















Emerging Disinfection By-Products, a growing concern

- Over 800 compounds
- Some of the emerging classes (haloacetonitriles, haloacetamides) are more toxic than regulated DBPs
- Iodinated-DBPs > Brominated DBPs > Chlorinated-DBPs

Wate

Not regulated, nor widely measured
 > Study of THMs as well as several other DBPs







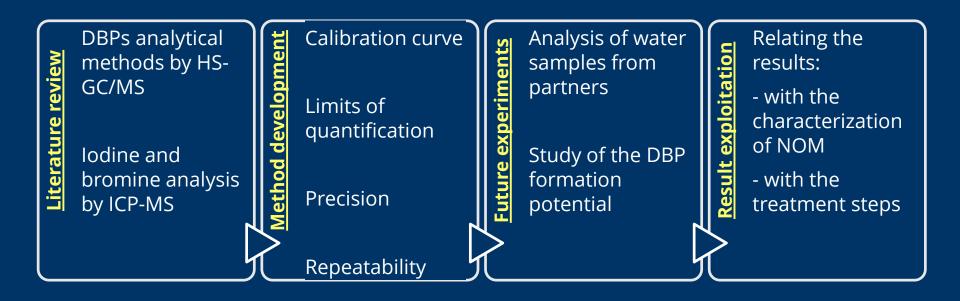






II. Research interest

Study plan

















II. Research interest

List of compounds

Trihalomethanes

• <u>Regulated</u>

- Chloroform
- Bromoform
- Bromodichloromethane
- Dibromochloromethane

• <u>lodinated</u>

- Iodoform
- Dichloroiodomethane
- Bromodiiodomethane
- Dibromoiodomethane
- Chlorodiiodomethane

Haloacetic Acids

• <u>HAA5</u>

- Monochloroacetic acid
- Dichloroacetic acid
- Trichloroacetic acid
- Monobromoacetic acid
- Dibromoacetic acid
- Iodoacetic acid

Other compounds

Haloacetonitriles

- Iodoacetonitrile
- Bromoacetonitrile
- Chloroacetonitrile

Haloacetamides

- 2-chloroacetamide
- 2-bromoacetamide
- 2,2-dichloroacetamide
- 2,2,2-trichloroacetamide











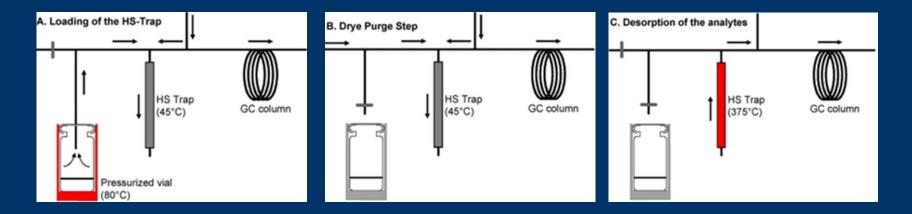




II. Research interest

Disinfection By-Product analysis by Headspace-Trap GC/MS

Less toxic solvents and lower limits of detection



Principle of sample preparation with headspace trap (Schulz K. et al. 2007)



TUDelft

III. Literature review

Important parameters to consider for HS-Trap GC/MS methods

Important parameters		Optimum conditions			
		THMs	HAAs	HANs	HAcAMs
Sample preparation	рН	2-9	5.2-7.1	2.6-5.6	4-7
	Organic modifier	none	n-pentane	MTBE/none	MTBE
	Derivatisation reagent	0	DMS + TBA–HSO4	0	0
HS-trap settings	Equilibration time (min)	10-30	20	20	NR
	Equilibration temp (°C)	60	60	80	NR
	Trap Desorb temp (°C)	200-250]180-230[for SPME	NR	NR















III. Literature review

Existing Headspace methods for simultaneous analysis

- Trihalomethanes and Haloacetic acids
 - Headspace method (Cardador et al. 2015)
- Haloacetonitriles and Trihalomethanes
 - Headspace methods (Antoniou et al. 2006; Montesinos and Gallego 2013; Luo et al. 2014)
- Haloacetonitriles and Haloacetamides
 - Liquid/Liquid extraction (*Bond et al. 2015*)









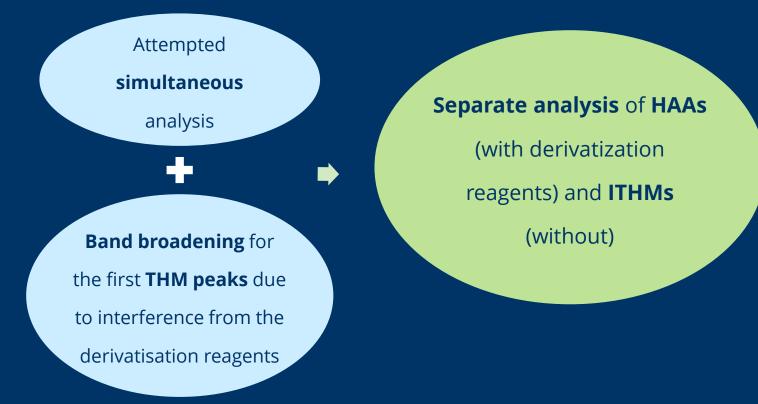






IV. Development of analytical methods

1. Analysis of HAAs and ITHMs separately

















IV. Development of analytical methods

2. Analysis of Haloacetonitriles and Haloacetamides

Haloacetamides L/L extraction with an organic solvent Haloacetonitriles by HS techniques with/without a small amount of an organic solvent

Analysis with the ITHMs
 If not possible, analysis

 of the remaining
 compounds separately
 with an organic solvent















IV. Development of analytical methods

3. Other analytical methods

lodide, iodate, bromide and Total Organic Iodine Total iodine analysis by **ICP-MS** and Bromine analysis by **HPLC-ICPMS** Fate of iodine and bromine during the different treatment steps









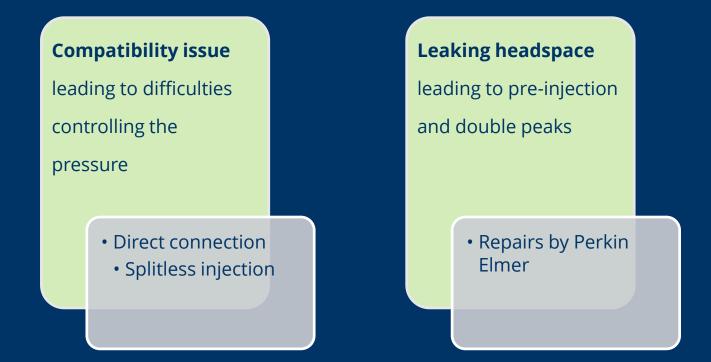






V. First results

1. Initial Headspace instrument problems









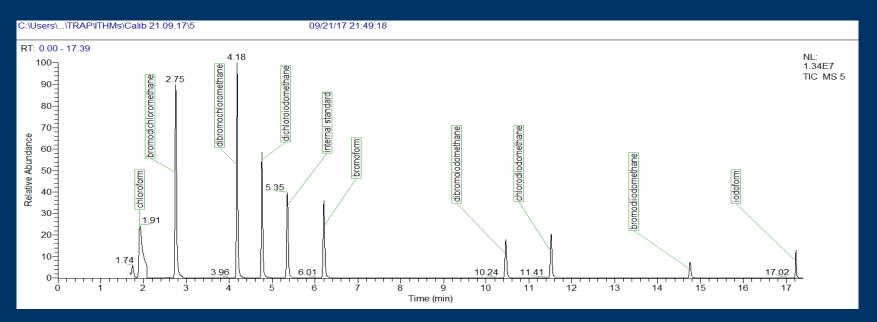






V. First results

2. Separate analysis of ITHMs



• Good peaks except for chloroform

Matrix issue for chloroform

- RSD < 10% for 5 standards at 5 μg/L
- LOQ < 50 ng/L















VI. Upcoming work

Other analytical methods

- Analysis of Haloacetic Acids, Haloacetonitriles and Haloacetamides
- Analysis of iodide, iodate, bromide and Total Organic Iodine and Bromine
- Analysis of real water samples to study the DBP formation potential (by testing disinfection scenarios)















VI. Upcoming work

Analysis of real samples

- 2 samples:
 - Raw water
 - Non-chlorinated treated water samples
- 1 L bottles filled to the rim
- Start: November 2017



To determine the best treatment to reduce DBP formation potential







Thank you for your attention

References

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